Application No. 10/660,687

Reply to Office Action

AMENDMENTS TO THE CLAIMS

This listing of claims replaces all prior versions, and listings, of claims in the application.

- 1. (Currently Amended) A <u>chemical-mechanical polishing composition</u> system for planarizing or polishing a composite substrate comprising (i) a polishing composition comprising (a) about 0.5 wt.% to about 10 wt.% of a source of fluoride ions, (b) about 1 wt.% to about 10 wt.% of an amine, (c) about 0.1 wt.% or more 0.5 wt.% to about 8 wt.% of a base, and (d) water, and (ii) about 0.1 wt.% to about 30 wt.% of an abrasive, and (e) water.
- 2. (Currently Amended) The <u>composition</u> system of claim 1, wherein the <u>composition</u> system has a pH of about 7-14.
- 3. (Currently Amended) The <u>composition system</u> of claim 1, wherein the abrasive is selected from the group consisting of alumina, silica, titania, ceria, zirconia, germania, magnesia, coformed products thereof, and mixtures thereof.
- 4. (Currently Amended) The <u>composition</u> system of claim 3, wherein the abrasive is silica.
 - 5. (Canceled)
 - 6. (Canceled)
- 7. (Currently Amended) The <u>composition</u> system of claim 1, wherein the fluoride ions are from a source of fluoride ions selected from the group consisting of fluoride salts, fluoride acids, fluoride metal complexes, and combinations thereof.
- 8. (Currently Amended) The <u>composition</u> system of claim 1, wherein the amine is an amino alcohol.
- 9. (Currently Amended) The <u>composition</u> system of claim 8, wherein the amine is 2-dimethylamino-2-methyl-1-propanol.
- 10. (Currently Amended) The <u>composition</u> system of claim 1, wherein the base is selected from the group consisting of inorganic hydroxide bases and carbonate bases.

Application No. 10/660,687

Reply to Office Action

- 11. (Currently Amended) The <u>composition system</u> of claim 10, wherein the base is selected from the group consisting of potassium hydroxide, sodium hydroxide, ammonium hydroxide, cesium hydroxide, sodium carbonate, and mixtures thereof.
- 12. (Currently Amended) The <u>composition</u> system of claim 1, wherein the <u>composition</u> system further comprises a quaternary ammonium compound.
- 13. (Currently Amended) The <u>composition</u> system of claim 1, wherein the <u>composition</u> system has a polishing selectivity of oxide:nitride of about 2:1 or more.
- 14. (Currently Amended) The <u>composition</u> system of claim 1, wherein the <u>composition</u> system comprises a cationic species that reduces nitride removal from the composite substrate.
- 15. (Currently Amended) The <u>composition</u> system of claim 1, wherein the fluoride ions comprise less than about 100% active fluoride ions.
- 16. (Currently Amended) The <u>composition</u> system of claim 1, wherein the <u>composition</u> system has a free alkalinity value of about 0.001-0.15 mol/l.
- 17. (Currently Amended) The <u>composition system</u> of claim 1, wherein the <u>composition system</u> has a total alkalimity value of about 0.005-0.2 mol/l.

18-42. (Canceled)